CARDIAC GLYCOSIDES: 3. SYNTHESIS OF 6-D-DIGITOXOSE ANALOGUES

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Abstract--Earlier studies have shown that the β -D-digitoxose (2,6-dideoxy- β -D-mannopyranose) directly attached to the cardiac glycoside steroid C3 has the greatest effect on biological activity. This report describes the synthesis of eight digitoxosides (2a-9a), with widely varying cyclic and acyclic C17 β -side groups, and the corresponding C3',C4'-acetonides (2b-9b). NMR analysis of conformational strain introduced by the acetonide groups is supported by crystallographic analysis of the sugars' torsion angles.

Cardiac glycosides² such as digoxin (1a)and digitoxin (1b) (Chart I) are currently among the six most-prescribed drugs in the U.S.³ Their structures include three g-Ddigitoxose (2.6-dideoxy-B-D-mannopyranose) sugars attached to the genin C3-0. Because of these glycosides' clinical and biochemical importance, we have been very interested in synthesizing new analogs and in delineating their pharmacological mechanisms of action. In studies with a variety of cardenolide genins, we found that the position of each side group's carbonyl oxygen (or in the case of a nitrile, its nitrogen) was a nearly perfect predictor of biological activity.⁴ Most recently, we have identified the C17 side group receptor subunit on Na+.K+-ATPase using a site specific photoaffinity probe.^{4f}

We have now turned our attention to a systematic study of the roles of structure and conformation of the C3 sugars; and in particular how they relate to the effects of modified C17 β side groups.^{1,5,6} The synthesis of

1) their natural occurance of β-D-digitoxose in digitoxin and digoxin; 2) the 'first' digitoxose on digitoxin (i.e., the one directly attached to the genin C3) increases activity relative to digitoxigenin ten-fold; but the 'second' and 'third' digitoxoses have very little additional effect; 6,7 and 3) the possible clinical applications of a 'weak' C17B side group combined with a sugar which increases binding and activity. The genins were selected because of the structural variety of C17ß side groups, and because they have over a 100-fold range in activities.⁴ The acetonides 2b-9b were also of interest because (in addition to being synthetic intermediates to 2a-9a) they could be used in future biological studies to evaluate the roles of the 3' and 4' hydroxyls.

β-D-digitoxides **2a-9a** was undertaken because,

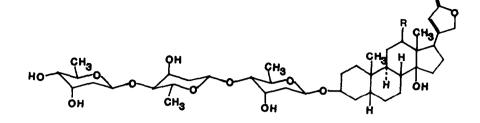
RESULTS AND DISCUSSION

Digitoxigenin β -D-digitoxoside (38, 58, 148)-3-[2,6-dideoxy- β -D-ribohexopyranosyl) oxy]-14-hydroxycard-20(22)-enolide (2a) is the central intermediate in the synthesis of all the other β -D-digitoxosides reported in this study. It has previously been prepared in

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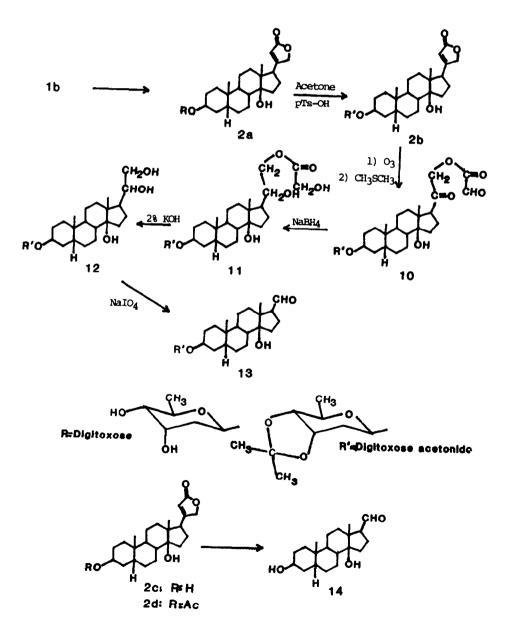
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1a: R= OH

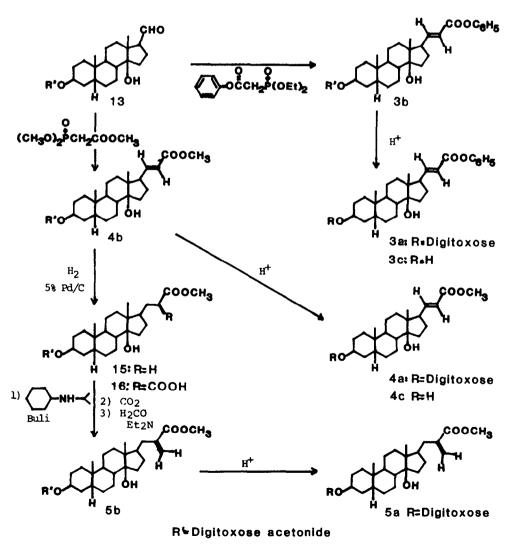
16: R. H



small (10%) vield by Zorbach^{8a} using acid catalyzed addition of digitoxose to digitoxigenin. Modified Koenigs-Knorr reaction of di-0-p-nitrcbenzovl digitoxose chloride with digitoxigenin was found by Zorbach^{8b} to give the a-D-digitoxoside in 45% yield but little or no 2a. The best yields of 2a have been reported by Satoh and Aoyama⁹ using a stepwise degradation of the two terminal digitoxoses of digitoxin (1b). The 3' and 4' hydroxyl groups of 2a were then protected (acetone and ptoluenesulfonic acid) to give acetonide 2b (36,56,146)-3-2,6-dideoxy-3,4-0-(1-methy]ethylidine-B-D-ribo-hexopyranosyl)oxy]-14hydroxycard-20(22)-enolide) (Scheme 1).

Aldehyde **13** (38,58,148,178)-3-[(2,6dideoxy-3,4-<u>0</u>-(1-methylethylidine)-8-D-ribohexopyranosyl)oxy]-14-hydroxyandrostane-17carboxaldehyde) was prepared using a modification of a procedure used by us^{10} and by Thomas¹¹ to convert digitoxigenin (2c) to its acetate 2d, and then to the genin aldehyde 14. Ozonolysis of 2b followed by reduction of the ozonide with dimethylsulfide instead of zinc and acetic acid provided dialdehyde 10, which was reduced with sodium borohydride to diol 11. The hydrolysis of 11 with methanolic potassium hydroxyide and oxidation of the product 12 with sodium periodate gave aldehyde 13 in 72% yield overall yield from 2b.

Wittig^{12,13} reaction of **13** (Scheme 2) with phenoxycarbonylmethyl methyl phosphonate (prepared from phenyl bromoacetate and trimethyl phosphite (see Experimental Section)

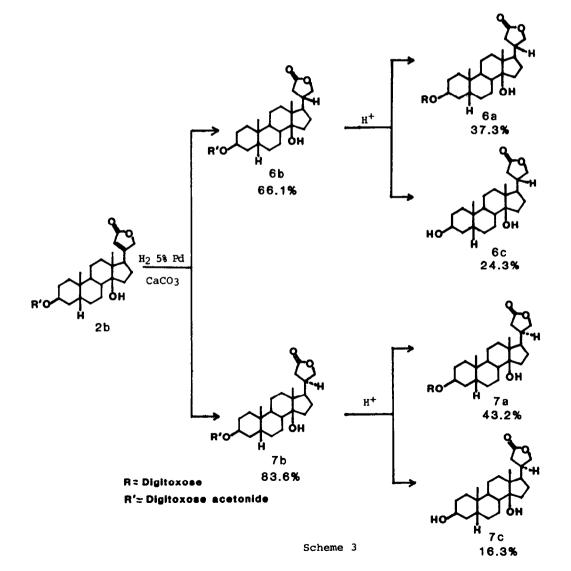


Scheme 2

gave the phenoxy β-D-digitoxoside acetonide 3b (38,58,148,20E)-pheny1-3-[(2,6-dideoxy-3,4-0-(1-methylethylidine)-β-D-ribo-hexopyranosyl) oxy]-14-hydroxy-(pregn-20-ene-21-carboxylate) in 51% yield. Reaction of 13 with trimethyl phosphonoacetate--a modification of the procedure used by Thomas¹¹ and ourselves¹ to convert digitoxigenin **2c** to genin methyl ester **4c--**gave the methyl ester β-D-digitoxoside acetonide 4b (38,58,148, 20E)-methyl 3-[2,6dideoxy-3,4-0-(1-methylethylidine)-&-D-ribohexopyranosyl-oxy]-14-hydroxyl-(pregn-20-ene-21-carboxylate) in 84% yield. Hydrolysis of the acetonide protecting groups (methanol. room temperature, trace of 5% HCl) invariably gave a small amount of the corresponding genin and unreacted acetonide as well as the desired B-D-digitoxoside. Hydrolysis of 4b thus gave

4a (48.2%) and 4c (trace), along with unreacted 4b; and 3b gave 3a (53.7%) and 3c (30.4%) (Scheme 2). The PMR coupling constants (16 Hz) between the C20 and C21 protons of these compounds indicate the trans stereochemistry of the C20 double bond; and confirmed by x-ray crystallography of 4a, 4b, and $4c^{14}$. All physical and spectral data (Tables 1 and 2) are in agreement with the assigned structures.

Catalytic hydrogenation of **4b** with 5% paladium on carbon as the catalyst^{3e} gave dihydro compound **15**. The enolate of **15** was treated with anhydrous CO_2 , followed by refluxing of the crude product **16** in aqueous formaldehyde and diethylamine (as previously reported by us for genins^{3e}) to give acetonide **5b** (38,58,148,20E)-methyl-3-[(2,6-dideoxy-3,4-

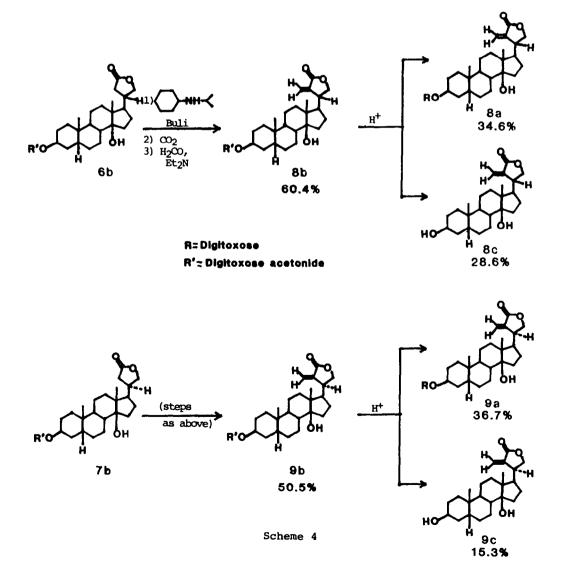


<u>0</u>-(1-methylethylidine)-β-D-ribo-hexopyranosyl)oxy]-14-hydroxy-21-methylene-(pregn-21-

carboxylate) in 30% overall yield. Hydrolysis of 5b gave 5a (35.7%) along with recovered acetonide 5b. We have previously reported that the (20R,S) mixture of genins (i.e., 6c and 7c) obtained by hydrogenation of digitoxigenin (2c) could be separated in good yield using fractional crystallization.^{3e,15} We were not able to thus separate B-D-digitoxosides **6a** and **7a** (Scheme 3) but we could separate acetonides 6b (20R) and 7b (20S), 83.6% and 66.5% respectively from dihydro product **2b**, using a combination of methanol and ethyl acetate as a recrystallization solvent. The micro-TLC system developed¹⁵ for the separation of the cardenolide diastereomers was also successful in separating **6a** and

7a, an essential aid in following the progress of the fractional crystallizations. The chemical shifts and coupling constants δ 3.87(t, J=8 Hz) and δ 4.49(t, J=8 Hz) of the C21 protons of **6b** are clearly different from those of 7b, δ 4.02(t, J=9 Hz) and δ 4.41(t, J=9 Hz) (Table 3). No trace of **7b** could be seen in the TLC nor in the PMR spectrum of 6b; nor vice-versa for 6b, providing good proof of diastereomeric purity. Absolute structural assignment of stereochemistry at C20, however, was provided by comparison with an authentic sample of 7b and 7c whose x-ray crystallographic analysis have already been completed.4b,14

22-Methylenation of **6b** and **7b** (Scheme 4) was completed in the same way used for **4b**, giving acetonides **8b** and **9b**, 50.5% and 60.4%



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overall yields, respectively. Subsequent hydrolysis gave g-D-digitoxoside 8a and 9a. along with the corresponding genins **8c** and 9c. Absolute proof of C20 stereochemistry was on the basis of spectral comparison with our sythentic samples of **8c** and **9c^{3e}--**and an x-ray crystal structure has been completed on the authentic sample of 8c and 9b.4b,14 Although the PMR and ¹³C-NMR spectra of some other digitoxigenin mono-alycosides have been reported^{11b,17,18} they have not been for digitoxosides nor their acetonides. The protons (Table 2) of the steroid skeleton were easily assigned by comparing chemical shifts and coupling constants with those of the corresponding genins.^{4e} Assignment of the sugar protons of the methyl ester acetonide 4b was completed by decoupling experiments. Irradiation of C6'-H (δ 1.16,d,J= 6Hz) gave a doublet (J=9 Hz) for C5'-H (δ 3.44), and octet before decoupling. The octet (J=9 and 6 Hz) of C5'-H was irradiated to collapse C6'-H to a singlet. The irradiation of C2'-H (§ 2.07) gave a doublet (J3'-4'=5 Hz) for C3'-H ($_{\delta}$ 4.40) and singlet Cl'-H (§ 4.77). Irradiation of C3'-H (δ 4.40) produced a doublet (J=9 Hz) for C4'-H The assignment of the PMR spectra (& 3.76). for the other compounds' sugars was then made on the basis of those of 4b (see Table 2). The conditions used during the synthesis did

not affect the anomeric (C1') β stereochemistry from the original starting material digitoxin (**2b**). The C1' proton of each digitoxoside has a doublet of doublets PMR absorption, J=9 or 8 Hz and J=2 or 3 Hz. This absorption is in good agreement with the natural digitoxoside **2a**. Although we would not expect a change in the anomeric carbon under the conditions used in the work, it should be noted that if α -anomers were present, the C1' coupling constants should be 3 Hz or less.¹⁹

It may be noted in Table 2 that the C1' protons in the acetonides are at higher field than those for the free digitoxosides. The coupling constants (J=6 Hz) between C3'-H and C4'-H in the acetonides are also larger than that (J=5 Hz) of the free digitoxosides. These data suggest that the pyranose rings in the acetonides aredistorted by the protecting group itself.

The crystal structures of five digitoxosides ($4a^{14}$, gitoxin,²⁰ digoxin,²¹ and the mono²² and digitoxosides²³ of digoxin) and five digitoxose acetonides (2b,4b,5b,7b, and 9b¹⁴) provide the structural data to substantiate the conclusions drawn from the spectral results. The averaged structural parameters for the sugar molety in each group show a high degree of internal consistency. Comparison of the corresponding averaged parameters for the

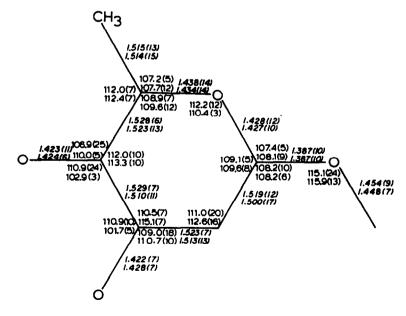


Figure 1

Average bond lengths and bond angles for the digitoxoside portion of cardiac glycoside analogs containing a digitoxoside, top values, or a digitoxose acetonide, bottom values. The numbers in parentheses are the standard deviation of the average.

two groups shows that there are no significant differences in the bond lengths, but that there are four bond angles which show significant variations: C2'-C3'-C4', C4'-C3'-03', C3'-C4'-04' and C5'-C4'-04'. However, only the C2'-C3'-C4' angle is in the hexopyranosyl ring, the other three angles are all exocyclic (see Fig. 1).

The conformation of the hexopyransoyl ring is significantly flattened by the acetonide group. The C1'-C2'-C3'-4' and C2'-C3'-C4'-C5' torsion angles are distorted by 12.5° and -13.4° respectively in going from the free digitoxoside to the digitoxose acetonide. All of the torsion angles involving 03' and 04' are also significantly altered by the presence of the acetonide, see Table 3. Thus, the spectral results and the crystal structure data are in perfect agreement.

Hog kidney Na+,K+-ATPase inhibitory studies have been completed on the digitoxosides and acetonides described in this paper.¹⁴ The digitoxosides were consistantly about an order of magnitude more active than the corresponding genins, but the acetonides were only about twice as active as the genins. The glycosides carbonyl oxygen positions were also shown to correlate very well with those of the genins. The contributions of the sugars to Na+,K+-ATPase inhibitory activity is quite separate, confirming the proposal first made by Yoda that the Na+,K+-ATPase receptor has distinct sugar and sugar binding sites.

	·	ELEMENTA	L ANALYSI	S IR SPECT	RA (cm-l):	KBr	[a] D ²² (c)
	FORMULA	CALČD. (FOUND)	CALCD. (FOUND)	он	C=0	C=C	in MeOH
2a	C29H4407	69.02 (69.35)	(9.00)	3450 (br)	1725(br) 1775(sh)	1620	-5.6°a)(0.305)
2b	C32H4807	70.56 (70.60)	(9.00)	3500	1740	1625	+8.5°(0.270)
36	C37H52O7	73.00 (73.81)	(8.61)	3440 (br)	1720	1640	+27.7°(0.300)
4b	C32H5007	70.30 (70.12)	9.22 (8.96)	3250 (br)	1710	1650	+18.7°(0.300)
3a	C34H4807	71.80 (71.68)	8.51 (8.78)	3400 (br)	1720	1640	+13.7°(0.300)
4 a	C29H4607	68.75 (68.54)	9.15 (9.05)	3550 3420(bd)	1720	1640	+10.0°(0.300)
5b	C33H5207	(70.68 (70.83)	(9.25 (9.37)	3500(br)	1720	1630	+10.0°(0.310)
5a	C30H4807	(69.20) (69.26)	9.29 (9.30)	3440(br)	1710	1630	-2.6°(0.350)
6a	C29H4607	68.75 (68.55)	9.15 (9.00)	3480(br)	1770		-4.3°(0.305)
7a	C29H4607	68.75 (69.00)	9.15 (8.89)	3500(br) 3380(br)	1775(sh) 1750		-10.6°(0.265)
6b	C34H5007	70.30 (70.10)	9.22 (0.29)	3500(br)	1775		+12.1°(0.214)
7b	C32H5007	70.30 (70.57)	9.22 (8.95)	3520	1780 1760		+2.6°(0.232)
8b	C33H5007	70.94 (71.11)	9.02 (8.94)	3510	1750	1656	-49.5°(0.275)
9b	C33H5007	70.94 (71.15)	9.02 (8.93)	3490	1750	1660	-79.30(0.275)
8a	C30H4607	69.47 (69.62)	8.94 (8.74)	3440(br)	1750	1660	-64.8°(0.290)
9a	C30H4607	69.47 (69.43)	8.94 (8.92)	3500(br)	1750	1660	+69.5°(0.220)
13	C29H46O6	70.99 (70.92)	9.45 (9.65)	3470	1705		-14.7°(0.300)
15	C 32H 52O 7	70.04 (70.21)	9.55 (9.60)				+14.8°(0.330)
<u>_</u>)	D Satoh an		reported	-5.2% (c	- 0.327 N	еон)]	

TABLE 1. PHYSICAL DATA OF DIGITOXOSIDES

a) D. Satoh and K. Aoyama reported -5.2° (c = 0.327, MeOH)1

EXPERIMENTAL

Melting points were determined with a Thomas-Hoover melting point apparatus and are uncorrected. Elemental analyses were performed by MHW Laboratories, Phoenix, Ariz. The 100-MHz PMR were taken in CDC1₃ at the Oregon State University Department of Chemistry Spectroscopy Laboratory. IR spectra were run as KBr pellets or NaC1 films (for liquids) using a Beckmann Model Acculab 7 spectrophotometer. Optical rotations in methanol were taken on a Perkin-Elmer 141 polarimeter. Ozonolysis was performed using the Ozonator Model O3V10-0 from the Ozone Research and Experiment Corporation (OREC).

Research and Experiment Corporation (OREC). Dry column chromatogrpahies' (as developed by K. Yoshioka and coworkers at Takeda Research Laboraories but unpublished, and modified by us) were completed using EM silica gel 60 (70-230 mesh) in a Buchner funnel with medium porosity fritted disk, 15 ml funnel for 50-100 mg compound; 60 ml funnel for 100-500 mg compound; 150 ml funnel for 500 mg - 2 g compound; with 10, 30, and 50 g silica gel respectively. The compound (or crude product) is dissolved in appropriate solvent (usually methylene chloride) having the same volume as the amount of silica gel used. (In column chromatography, the goal is normally to dissolve the compound in as little solvent as possible.) Subsequent elution of compounds was with the same solvent system used for thin layer chromatography (described below).

Thin layer chromatographies were on microscope slide sized .25 mm EM silica gel 60 F-254 as previously described.¹⁵ Solvent combinations used were: "A" CH₂Cl₂ 20: EtOAc 2: MeOH 1; "B" CH₂Cl₂ 10: EtOAc 3: MeOH 1. Preparative Chromatographies were on plates coated by us with EM silica gel 60 PF-254 .75 mm. Analytical tlc plates were visualized by UV light at 254 nm, iodine vapor, or by spraying with 2% CeSO₄ in 2N H₂SO₄ and heating in a hood on a hot plate. Spectral, analytical, and optical rota-

Spectral, analytical, and optical rotation data for final compounds are shown in Tables 1 and 2.

(38,58,148)-3-[2,6-dideoxy-3.4-0-(1-

TABLE 2.	PMR	SPECTRA ^a , b)	ÛF	DIGITOXIDES	IN	CDC13	(δ)
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	С1'-Н	С2'-Н	С3'-Н	C4'-H	С5'-нс)	С6'-Н	С3-Н	С20-Н С18-Н ^д)	C19-Hd
2b	4.78 (dd,9,3)		4.42 (m)	3.68 (dd,8,5)	3.44 (m,8,6)	1.24 (d,6)	4.03 (m)	0.86	0.92
2a	4.89 (dd,9,3)		4.12 (m)	3.73 (dd,9,6)		1.28 (d,6)	4.04 (m)	0.88	0.93
3b	4.79 (dd,8,3)		4.42 (m)	3.68 (dd,9,5)	3.46 (m)	1.24 (d,6)	4.07 (m)	0.19	0.92
3a	4.90 (dd,9,3)		4.12 (m)	3.73 (dd,9,5)	3.37 (m)	1.25 (d,6)	4.08 (m)	7.28 0.90 (dd,16,10)	0.92
4b	4.77 (dd,9,3)	2.07(m) 2.18-1.98	4.40 (m)	3.67 (dd,9,5)	3.44 (m,9,6)	1.16 (d,6)	4.03 (m)	7.17 0.87 (dd,16,10)	0,93
4a	4.88 (dd,9,3)	2.02(m)	4.11 (m) ¹	3.66 (dd,9,6)	3.45 (m,9,6)	1.29 (d,6)	4.02 (m)	7.18 (dd,16,10) ^{0.85}	0.92
5a	4.88 (dd,8,2)		4.11 (m) ¹	3.66 (dd,9,6)	3.37 (m,9,6)	1.28 (d,6)	4.02 (m)	0.93	0.98
5b	4.77 (dd,8,3)		4.43 (m)	3.65 (dd,9,5)	3.45 (m,9,6)	1.26 (0,6)	4.03 (m)	0.95	1.00
6b	4.76 (dd,8,3)		4.41 (m)	3.66 (dd,9,5)	3.44 (m,9,6)	1.24 (d,6)	4.01 (m)	0.94	0.94
6a	4.86 (dd,8,2)		4.09 (m)	3.71 (dd,9,6)	3.33 (m,9,6)	1.27 (d,6)	4.02 (m)	0.91	0.93
7b	4.76 (dd,8,3)		4.39 (m)	3.66 (dd,9,5)	3.43 (m,9,6)	1.24 (d,6)	3.99 (m)	0.92	0.96
7a	4.86 (dd,8,3)		4.11 (m)	3.72 (dd,9,6)	3.36 (m,9,6)	1.28 (d,6)	4.02 (m)	0.92	0.97
8b	4.75 (dd,9,3)		4.39 (m)	3.66 (dd,9,5)	3.42 (m,9,6)	1.23 (d,6)	4.00 (m)	3.25 0.92 (m)	0.98
8a	4.75 (dd,9,2)		4.13 (m)	3.73 (dd,9,6)	3.37 (m,9,6)	1.28 (d,6)	4.02 (m)	3.27 0.93 (m)	0.99
9b	4.77 (dd,9,3)		4.41 (m)	3.67 (dd,9,5)	3.44 (m,9,6)	1.23 (d,6)	4.01 (m)	3.29 0.90 (m)	0.91
9a	4.87 (dd,9,2)		4.12 (m) ²	3.73 (dd,9,6)	3.36 (m)	1.28 (d,6)	4.03 (m)	3.28 0.90 (m)	0.91
13	4.73 (dd,9,3)		4.37 (m)	3.60 (dd,9,5)	3.39 (m)	1.16 (d,3)	3.97	9.7 0.93	0.85
15	(dd,8,3)		4.41 (m)	3.66 (dd,9,5)	3.44 (m,9,6)	1.29 (d,6)	4.01 (m)	0.92	0.95

a) m=multiplet, dd=doublet of doublets.
 b) Coupling constants in Hz are in ().

c) Each is an octet.
 d) Singlet

methylethylidine)&-D-ribo-hexopyranosyl)oxy]-14-hydroxycard-20(22)-enolide (2D). Digitoxigenin monodigitoxoside mp 200-202° (2a) was prepared from digitoxin using the method of Satoh. It (812 mg, 1.6 mmol) was then dissolved in acetone (30 ml). To this solution was added 8 mg of paratoluenesulfonic acid with stirring at room temperature for 3 hr. The reaction mixture was concentrated to half the volume in vacuo at 50°C. Water was added to the residue and this was extracted three times with CH₂Cl₂. The CH₂Cl₂ layer was washed with 5% NaRCO₃ and with water. After drying over MgSO₄, the solvent was evaporated in vacuo to afford 900 mg of crude oil of 3b, which was purified using dry column chromatography and preparative TLC using solvent system A. The main product, Rf = 0.65, was extracted with CH₂Cl₂ - MeOH - acetone (1:1:1) to give 806 mg of pure oil which was crystallized by acetone-hexane to afford 650 mg (74%) of 2b, mp 202-206°.

(38,58,148,178)-3-[(2,6-dideoxy-3,4-0-(1methylethylidine-β-D-ribo-hexopyranosyl)oxy]-14-hydroxyandrostane-17-carboxaldehyde (13).

TABLE 2 CONTINUED

Acetonide **2b** (2 g, 3.67 mmole) was dissolved in 100 ml of CH_2Cl_2 and cooled at -78° with a dry-ice bath. This solution was ozonized for 5 hr to afford a deep purple color characteristic of the ozonide. After removing of excess ozone by flushing with nitrogen, to the clear ozonide solution was gently added 4 ml of dimethylsufide and allowed to stand overnight raising from -78° to room temperature as the dry-ice sublimed. The reaction mixture was washed three times with H₂O. The methylene chloride layer was dried over MgSO₄, filtered and evaporated in vacuo to yield a crude powder of **10** (2 g), which was shown on TLC using solvent system B to be a single spot. The crude **10** was then dissolved in 100 ml of MeOH. To this solution was slowly added 316 mg (8.35 mmol) of NaBH₄ and stirred at room temperature for 1 hr. The reaction mixture was evaporated in vacuo to half the volume, H₂O was added and the mixture was extracted three times with ether. The ether extract was washed with water and with saturated NaCl, dried over MgSO₄, and evaporated in vacuo to 2.46 g of crude oil **11**. This was

	На	С21- НЬ	С22-Н	CH2 =	CH3 of <u>Acetonide</u> d) of COOCH3
2b	5.04 (dd,18,2)	4.78 (dd,18	5.88(m) ,2) Wh/2_3Hz		
2a	5.03 (d,18)	4.79 (d,18			
3b		5.00 (d,16)			1.45,1.33
3a		5.89 (d,16)			
4b		5.62 (d,16)	······		<u>1.58,1.36</u> 3.71
4a		5.63 (d,16)	······································		3.70
5a				6.12(m,Wh/2 3Hz) 5.53(m,Wh/2 3Hz)	3.73
56				6.12(m,Wh/2 3Hz) 5.53(m,Wh/2 4Hz)	<u>1.49,1.36</u> 3.75
6b	3.87 (t,8)	4.49 (t,6)		1.46,1.34
6a	3.87 (t,8)	4.49 (t,8) 3)		
7b	4.02 (t,9)	4.41 (t,9			1.46,1.34
7a	4.03 (t,9)	4.41 (t,9	; ;		
86	4.18 (dd,9,3)	4.39 (dd,9,	7)	6.26(d,2) 5.69(d,2)	
8 a	4.19 (dd,9,3)	4.41 (dd,9,		6.26(d,2) 5.72(d,2)	1.46,1.34
9b	4.35 (dd,10,8)	4.70 (dd,10		6.25(d,3) 5.53(d,3)	
9a	4.36 (dd,10,8)	4.70 (dd,10		6.28(d,3) 5.52(d,3)	
13	<u></u>	<u> </u>	- <u>-</u>		1.39,1.27
15			······		1.47,1.35
					3.75

shown on TLC in solvent system B to have two close spots due to the epimers at C20. The crude oil of 11 was dissolved in a solution of 2% KOH/MeOH (100 ml) and refluxed for 30 min on a steam bath. After cooling to room temperature, the reaction mixture was roto-evaporated to half the volume and the pH of the solution was adjusted between 5-6 with glacial acetic acid. Water was added slowly and the crystals formed were filterated after cooling in an ice bath. The crude crystals were washed with water, dried under vacuum to afford 1.4 g of 12. To a solution of 12 (1.4 g) in MeOH (60 ml) was added NaIO₄ (1.4 g) in H₂O (14 ml) and stirred at room temperature for 1 hr. The salt of the NaIO₃ formed was filtered and washed with MeOH. The methanol filtrate was concentrated to half the volume in vacuo and water was added to the residue. The milky solution thus obtained was cooled in ice bath. The resulting crystals were collected by suction, washed with water and dried to afford 1.3g (72%) of 13, mp 75-80°. (38,58,148,208)-phenyl-3-[2,6-dideoxy-

(3β,5β,14β,20E)-pheny1-3-[2,6-dideoxy-3,4-0-(1-methyl-ethylidine)-B-D-ribo-hexopyranosyl)oxy]-14-hydroxy-(pregn-20-ene-21carboxylate) (3b).

a) Synthesis of phenoxycarbonylmethyl diethyl phosphonate. Phenyl bromoacetate: Bromoacetyl bromide (50 g) was added dropwide (30 min) to an ethanol solution (400 ml) of phenol (23.5g) and pyridine (22 g), and the solution was stirred in a water bath for 1 hr at room temperature. After the pyrinine-HBr salt was filtered off, the filtrate was concentrated in vacuo to give a crude oil which was separated by dry column chromatography using solvent system A. The product with Rf 0.85 was distilled under reduced pressure to yield 13.5g (25%) of phenyl bromoacetate, bp 134-138°/15 mm Hg. After cooling at -20°, the oil gave crystals, mp 32°, IR(NaCl film) 1750 (C=0), 1590 and 149° (aromatic) cm⁻¹; NMR (CDCl₃, 60 MHz) δ 4.0 (s, 2H, COCH₂Br), 7.2 (m, 5H, aromatic). Synthesis of the phosphonate: Phenyl bromoacetate (13 g, 60.45 mmol) was mixed with 14.1 g (84.86 mmol) of triethyl phosphite and refluxed at 155° on an oil bath for 1 hr with generation of ethyl bromide. The crude oil was distilled to yield 14.4 g of pure phosphonate, bp 150-153°/.2 mmHg. (Lit¹ bp 174-176° 1.25 mmHg). IR (NaCl): 1750 (C=0), 1595, 1500 (aromatic), 1250(-PO-) cm⁻¹; NMR (CDCl₃, 60 mHz) δ : 1.38 (t, 3H, OCH₂CH₃), 3.18 (d, J=23Hz, -COCH₂PO-) 4.13 (m, 2H, OCH₂CH₃), 7.25 (m, 5H, aromatic).

(m, 5H, aromatic). **b)** Synthesis of '3b: To sodium hydride (50% in oil, 9.8 mg, 2.03 mmole, washed twice with hexane) was added 3 ml of dry diglyme. The phosphonate (520 mg, 2.03 mmol) in dry diglyme (5 ml) was then added at room temperature under nitrogen. After stirring for 15 min, 13 (400 mg. 0.81 mmol) dissolved in 5 ml of dry diglyme was added and stirred for 1.5 hr. The reaction mixture was diluted with large excess of acetate buffer (pH 5-6), then cooled in ice. The crystals were filtered and dissolved in CH_2Cl_2 , dried over $MgSO_4$, filtered, and evaporated to dryness in vacuo to yield a crude oil (504 mg), which was separated by dry column chromatography with solvent system A to give 305 mg of pure oil. This was converted to a powder by 'crystallization' in hexane, to yield 250 mg (51%) of **3b**, mp: 82-84.

 $(3\beta,5\beta,14\beta,20E)$ -phenyl-3-[(2,6-dideoxy-B-D-ribo-hexopyranosyl)oxy]-14-hydroxypregn-20ene-21-carboxylate) (3a). To a solution of 3b (160 mg, 0.262 mmol) in MeOH (50 ml) was added 2 drops of 5% HCl. This was stirred at room temperature for 6 hr. The reaction mixture was poured into water and extracted three times with CH₂Cl₂. The methylene chloride layer was washed with water, dried over MgSO₄, and evaporated in vacuo to give an oil. The crude oil was separated by dry column chromatography and preparative TLC to afford 41 mg (25.6%) of starting material 3b from the band of Rf 0.94. The extraction of Rf 0.49 gave 35

Table 3: AVERAGE TORSION ANGLES OF DIGITOXOSIDES VS ACETONIDES

Ave. Torsion Angles in Degrees (standard deviation)

Atoms	Digitoxoside	Acetonide	Change
C1'-C2'-C3'-C4'	-50.1(2.0)	-37.5(2.2)	12.6
C2'-C3'-C4'-C5'	49.0(1.2)	35.6(2.2)	-13.4
C3'-C4'-C5'-05'	-54.3(2.5)	-48.3(4.7)	6.0
C4'-C5'-05'-C1'	63.1(2.2)	65.7(2.4)	2.6
C5'-05'-C1'-C2'	-65.1(3.2)	-68.2(2.5)	-3.1
05'-C1'-C2'-C3'	57.4(3.8)	52.5(4.6)	-4.9
03-C1'-C2'-C3'	174.0(3.6)	170.1(5.3)	-3.9
03-C1'-05'-C5'	177.8(3.7)	174.1(3.0)	-3.7
03'-C3'-C2'-C1'	72.0(1.0)	77.1(2.8)	5.1
03'-C3'-C4'-C5'	-72.0(2.3)	-84.1(2.5)	-12.1
03'-C3'-C4'-04'	47.3(2.7)	34.6(2.4)	-12.7
04'-C4'-C3'-C2'	168.3(2.4)	154.3(2.7)	-14.0
04'-C4'-C5'-05'	-175.9(2.5)	-162.8(4.4)	13.1
04'-C4'-C5'-C6'	65.6(2.2)	77.5(5.2)	11.9
C6'-C5'-C4'-C3'	-172.7(2.4)	-168.0(5.6)	4.7
C6'-C5'-05'-C1'	-175.4(2.5)	-171.8(3.0)	3.6

mg (30.4%) of genin **3c**. The main product **3a** (80 mg, 53.7%, mp 103-105) was obtained from the band at rf 0.37 as a powder and crystallization from hexane.

(38,58,148,20E)-methy1-3-[2,6-dideoxy-3,4-0-(1-methylethylidine)-B-D-ribo-heropyranosyl)oxy]-14-hydroxy-(pregn-20-ene-21carboxylate (4b). Methyl ester 4b was synthesized from 13 (245 mg. 0.5 mmole) in diglyme 2.5 ml, trimethylphosphnoacetate (228 mg, 1.25 mmole) and NarH (60 mg of 50% oil, 1.25 mmole) in diglyme 1.5 ml in a similar manner to that of 3b described above. The crude product was white needles of **4b** (230 mg, 83.9%) mp 147-149°.

38,58,148,20E)-methyl-3-[(2,6-dideoxy-B-D-ribo-hexopyranosyl)oxy]-14-hydroxypregn-20ene-21-carboxylate (4a). A solution of 4b (125 mg) in 100 ml of MeOH and 2 drops of 5% HCl was stirred for 5 hr. The reaction mix-ture was treated in a same way as that of 3bto afford crude product of 4a, which was purified by dry column chromatography using EtOAc-CH_2Cl_2 (9:1 to 2:8). The starting material 4b (34 mg, 27.2%, mp 147-149°) was recovered from first fraction. The second fraction gave 72 mg (62.1%) of 4a as an oil, which was recrystallized from EtOAc hexane to yield pure crystals (56 mg, 48.2%), mp 191-193°. A trace of genin **4c** which was also seen on TLC, same Rf as 4c we have previously reported, but it was not isolated.

(38,58,148,178)-methyl-3-[(2,6-dideoxy-3,4-0-(1-umethylethylidine)-B-D-ribo-hexopyranosyl)oxy]-14-hydroxy-(pregn-21-carboxylate (15). The unsaturated ester 4b (541 mg) in MeOH 100 ml was hydrogenated (1 atm) with in Media 100 mi was hydrogenated (1 atm) with 5% Pd-C (100 mg) for 5 hr at room tempera-ture. The filtration and evaporation of the reaction mixture afforded a powder (563 mg). This was recrystallized from CH₂Cl₂-hexane to give 425 mg (78.2%) of 15, mp 135-135°. (38,58,148)-methyl-3-[(2,6-dideoxy-3,4-0-(1 extintion of the hermometer)]

(1-methylethylidine)-B-D-ribo-hexopyranosyl) oxy]-14-hydroxy-21-methylens-(pregn-21-carboxylate) (5b). To an ice cooled solution of Nisopropylcyclohexylamine 0.5 ml (0.486 g, 277 mmol) in anhydrous THF 2.5 ml was added 1.7 ml (2.72 mmol) of 1.6 M butyl lithium in hexane under N_2 . The solution was stirred for 10 min and then at -78° for 20 min. To the solution and then at -78° for 20 min. To the solution was added 200 mg (0.364 mmol) of **15** in anhydrous THF 2.5 ml and stirred for 40 min. Anhydrous CO₂ (Matheson, Bone-Dry grade) was bubbled into the solution for 4 min and the Bubbling continued dry-ice bath was removed. for 25 min, during which time the solution warmed to room temperature. The reaction mix-ture was poured into 15 ml of 5% HC1 and extracted three times with ether, washed with water, dried over MgSO4, and evaporated to give a crude product, presumably 16. To this crude product was added 3 ml of 37% formalin (with 10% MeOH) and 0.8 ml of diethylamine. The solution was heated on a steam bath for 30 The reaction mixture was cooled in an min. ice bath and to it was added 1 ml of 5% HC1 and 9 ml of water. The mixture was extracted three times with ether. The ether layer was washed with water, dried over MgSO₄, and evaporated in vacuo to afford an oil. The oil was purified by dry column chromatography using CH₂Cl₂-AcOEt and recrystallized from CH₂Cl₂-hexane to give 61 mg (29.9%) of **5b**, mp 165-167°.

38,58,148,20E)-methyl-3-[(2,6-dideoxy-8-D-ribo-hexopyranosyl)oxy]-14-hydroxy-21methylene-(pregn-21-carboxylate) (5a). A solution of 5b (75 mg) in MeOH 50 ml was stirred with one drop of 2% HCl at room temperature for 75 hr. The reaction mixture was treated in the same way as that of 3b to give a powder (58 mg), which was purified by preparative TLC (solvent system B). The oily product (33 mg) obtained from the band at Rf 0.50-0.56 was recrystallized from EtoAc-hexane to yield 25 mg (35.7%) of **5a**, mp 172-174°. Starting material **5b** (10 mg, mp 165-166°) was recovered from the band Rf 0.86-0.95.

(20R and 205)-20,22-dihydrodigitoxigenin-3-[2,6-dideoxy-3,4-0-(1-methylethylidine)-B-Dribo-hexopyranosyl] (20R 6b and 20S 7b). To a suspension of pre-hydrogenated 5% Pd-CaCO₃ in MeOH (40 ml) was added **2b** (2 g, 3.67 mmol) and stirred at room temperature for 24 hr, as pre-viously reported.⁸⁰ New prehydrogenated cataviously reported.³⁶ New prehydrogenated cata-lyst (800 mg) was added to the reaction mixture and stirred for another 24 hr. The mixture was filtered and the residue was washed with methanol and combined filterate was concentrated to dryness in vacuo. the crude product was crystallized from acetone-hexane to give 1.45 g (72%) of a mixturgeoff **6b** and **7b**, as a powder, mp 186-188; $[\alpha]_{P}^{P} = +5.90$ (c=.305); IR(KBr) γ max 3500 (0H), 1770 (c=0) cm⁻¹; Anal (C₃₂H₅₀O₇) C.H. A 3.75 X 10⁻³ M solution of the mixture of **6b** and **7b** did not show any UV absorption at 217 nm, thus showing the absence of any unreacted 20(22)-ene starting material. The two diastereomers were separated cleanly by microscope slide TLC, using three developments (solvent System "A") as we have reported for separation of the diastereometric genins.¹⁵ Rf values were .76 for 20R-**6b** and .82 for 20S-**7b**.

a) Separation of **6b** and **7b**. (A typical separation followed by TLC as described above, began with a larger amount of **2b** than above.) Acetonide **2b** (4.5 g) was hydrogenated with 2.4 g of preactivated 5% Pd-CaCO₃ in 100 m) of MeOH as above. The reaction mixture was filtered by suction. The residue was washed with hot MeOH (130 ml X 3). After cooling the MeOH solution at room temp., 2.0 g of "I" was obtained by filtration. Recrystallization from MeOH four times gave 1.18 g of **6b** (mp 205-212°). The mother liquor separated from "I" was concentrated to dryness (1.3 g), and "I" was concentrated to dryness (1.3 g), and then recrystallized from EtOAc twice, giving crystals (763 mg, "II") enriched in 7b. The filtrate from the original reaction mixture was evaporated in vacuo to afford 1.0 g of residue "III", and recrystallized from EtOAc twice to yield 448 mg enriched in 7b. To this 448 mg was added "II" and followed by two more recrystallizations from EtOAc to yield 925 mg of 205-7b (mp 197-201).

Multiple recrystallization of the mother liquor from "I,II,III" from a combination of MeOH and EtOAc as above afforded 707 mg (mp 207-212°) of **6b** (total 1889 mg, 83.6% based on approximately 1:1 mixture of R and S in 22) and 578 mg (mp 196.5-201°) of 7b (total 1494 mg, 66.1%).

(20R)-20,22-Dihydrodigitoxigenin-3-(2,6-

 $dideoxy-\beta-D-ribo-hexopyranoside)$ (**5a**). A solution of **6b** (383 mg) in MeOH 200 ml was stirred with four drops of 5% HC1 at room tem-perature for 18.5 hr. Work up in a similar manner as that of **3b** gave a crude mixture (245 mainter as that of 30 gave a character (24) may be a separated by dry column chroma-tography using CH_2Cl_2 -EtOAc (9:1 to 3:7). The first fraction gave 29 mg (7.6%) of the start-ing material **6b** (mp 208-213°). The oily product (68 mg) from the second fraction was

recrystallized from EtOAc to 6 mg (16.1%) of recrystallized from EtOAc to 6 mg (16.1%) of genin **6a** (mp 211-212°), which we have previously reported^{3C}. The third fraction afforded a crude oil (196 mg) which was crystallized from EtOAc to 155 mg (43.7% based on recovered starting material) of **6a**, mp 220-221°.

(205)-20,22-Dihydrodigitoxigenin-3-(2,6 $dideoxy-\beta-D-ribo-hexopyranoside)$ (7a). Acetonide 7b (246 mg) was hydrolyzed with three drops of 5% HC1 in a same way as that of 6b to give recovered starting material 7b (13 mg, give recovered starting material 70 (13 mg, 5.3%, mp 197-199° from EtOAc); the corresponding and previously reported ³⁰ genin 7c (41 mg, 24.3%, mp 213-214° from EtOAc); and oily 7a (129 mg). The 7a thus obtained was recryst tallized with EtOAc to afford 85 mg (37.3%) of 7a. mp 211-214°.

22-Methylene-(205)-20,22-Dihydrodigitoxigenin-3-[2,6-dideoxy-3,4-0-(1-methylethylidine)-B-D-ribo-hexopyranoside] (8b). (20S)-Dihydro compound 6b (821 mg, 1.5 mmol) in anhydrous THF 15 ml was reacted with N-isopropylcyclohexylamine (1.8 ml, 1.74 g, 9.96 mmol) in anhydrous THF (9 ml) and 6.6 ml (9.9 mmol) of 1.5 M butyllithium in hexane followed by treating of the produced enolate with anhydrous CO_2 in a similar manner as that of **5b**. The isolated intermediate (933 mg), presumably the carboxylic acid, was heated with 10 ml of 37% formalin (10% MeOH) and 3 ml of diethylamine on a steam bath for 40 min. Work up the reaction mixture gave crude product (800 mg) which was purified by dry column chromatography using CH_2Cl_2 -EtOAc (9:1 to 8:2). The oily product obtained was crystallized from EtOAc to give 424 mg (50.5%) of **8b**, mp 196-200°.

22-Methylene-(20R)-20,22-Dihydrodigitoxigenin-3-[2,6-dideoxy-3,4-0-(1-methylethyli-dine)-B-D-ribo-hexopyranoside] (9b). The 22-methylene compound 9b was synthesized from 7b (600 mg, 1.1 mmol), N-isopropyl-cyclohexyla-mine (1.3 ml, 1.26 g, 7.23 mmole) in anhydrous THF (6.5 ml), 1.5 M butyllithium in hexane (48 The (0.5 mi), 1.5 m butyfilterium in means (40 ml, 7.19 mmol), 37% formalin (4 ml) and diethylamine (1.1 ml) by the same method as described above. The crude oil (660 mg) thus prepared was purified by dry column chromatography as above to give a pure oil, crystallized from ether-hexane to give 370 mg (60.4%) (60.4%)of 9b, mp 179-182°.

22-Methylene-(205)-20,22-dihydroditoxi-

genin-3-(2,6-dideoxy-8-D-ribohexopyranoside) (8a). Hydrolysis of 8b (300 mg) in 200 ml of MeOH with 4 drops of 5% HCl for 12 hr produced 218 mg of crude oil in the same manner as that 218 mg of crude oil in the same manner as that of **4b**. The oily product was separated by dry column chromatography using CH_2Cl_2 -EtOAc (8:2 to 6:4) to afford 40 mg (13.3%) of starting material **8b** (mp 195-199°), 32 mg (15.3%) of the corresponding and previously reported³⁶ genin **8c** (mp 234-240°) and a crude powder of **8a**. This was recrystallized from EtOAc-MeOH-CH₂Cl₂ to give 116 mg (41.7%) of **8a**, mp 204-205° (dec.). The analytical sample of **8a** was obtained by recrystallization from MeOH. give obtained by recrystallization from MeOH, giv-ing 102 mg (36.7%) mp 223-228° (dec.).

22-Methylene-(20R)-20,22-Dihydrodigitoxi-

(9a). The (20R) diastereomer 9a (153 mg, 51.5%, mp 163-164° (decomp.) was prepared by hydrolysis of 320 mg of 9b in the same condition as that used for 8a, together with 27 mg (8.4%) of unreacted material **9b**, mp 179.5-181.5° and 63 mg (28.3%) of genin **9c**, mp 127-132° (powder). The analytical sample of **9a** was obtained by recrystallization from

AcOEt: yield 103 mg (34.6%), mp 193-194° (dec).

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REFERENCES

¹Paper 2 in this series: M. Kihara, K. Yoshioka, E. Kitatsuji, T. Hashimoto, and D.S. Fullerton, *Steroida*, in press.

Fullerton, Storoida, in press. ²For reviews, see ^aW.W. Zorbach, and K.V. Bhat, Adv. in Carbohydrate Chem. **21**, 273 (1976); D.S. Fullerton, "Cardiac Steroids" in "Wilson and Gisvold's Textbook of Organic Medicinal and Pharmaceutical Chemistry." (Doerge, R.F., Ed.) Lippincott, Philadelphia, PA, 1982, pp 712-724; T. Reichstein, and E. Weiss, Adv. in Carbohydrate Chem. **17**, 65 (1962); R.E. Thomas, in "Burger's Medicinal Chemistry," 4th ed, (M.E. Wolff, ed.) Wiley Interscience, New York, NY. 1981, pp 47-81; ^eK. Greeff, Ed. "Cardiac Glycosides," Vols I and II. Springer-Verlag, NY. 1981. and II. Springer-Verlag, NY. 1981.

³"Top 200 Products" in *Drug Store News*, March, 1983.

^{4a}K. Ahmed, D.S. Fullerton, T. Deffo, E. Kitatsuji, D.C. Rohrer, and A.H.L. From, J. Biol. Cham. 258, 8092 (1983); D.C. Rohrer, D.S. Fullerton, K. Yoshioka, E. Kitatsuji, K. Ahmed, and A.H.L. From, Acta Cryst B39, 272 (1983); D.S. Fullerton, K. Yoshioka, D.C. Rohrer, A.H.L. From, and K. Ahmed, Science 205, 917 (1979); D.C. Rohrer, D.S. Fullerton, K. Yoshioka, A.H.L. From, and K. Ahmed, fn "Computer Assisted Drug Design," (E.C. Olson, and R.E. Christoffersen, Eds.), American Chemical Society, 1979, pp 259-279; D.S. Fullerton, K. Yoshioka, D.C. Rohrer, A.H.L. From, and Chem. 22, 529 (1979); T. Deffo, D.S. Fullerton, et al. Biochemiatry, in press. ^{4a}K. Ahmed, D.S. Fullerton, T. Deffo, E.

⁵D.S. Fullerton, M. Kihara, T. Deffo, and E. Kitatsuji, *J. Med Chem.* (in press, paper 1 in this series).

⁶D.S. Fullerton, K. Yoshioka, D.C. Rohrer, A.H.L. From, and K. Ahmed, *Mol. Pharmacol.* 17, 43 (1980).

⁷S. Yoda, A.M. Sarrif, and A. Yoda, ibid., 647 (1975) and cited references.

8aw.w. Zorbach, N. Henderson, and Sejtara Saeki, J. Org. Chem. 29, 2016 (1964); W.W. Zorbach, and T.A. Payne, J. Amer. Chem. Soc.
82, 4979 (1960).

⁹D. Satoh, and K. Aoyama, *Chem. Pharm. Bull.* 18, 94 (1970).

¹⁰D.S. Fullerton, M.S. Pankaskie, K. Ahmed and A.H.L. From, J. Med. Chem. 19, 1330 (1976).

11a_{R.E.} Thomas, and J.S. Boutagy, Aust. J. Chem. 24, 2723 (1971), L. Brown, J. Boutagy, and R. Thomas, Aransim. Forsch. 31, 1059 (1981).

¹²W.S. Wadsworth, Jr., and W.D. Emmons, J. Am.

Chem. Soc., 83, 1733 (1961).

¹³J.S. Boutaguy, and R.E. Thomas, *Chem. Rev.* 74, 1 (1974).

14D.c. Rohrer, D.S. Fullerton, K. Ahmed and A.H.L. From, unpublished results, manuscript in preparation. (The refined X-ray co-ordinates have been sent to the Tetrahedron Editor.)

¹⁵K. Yoshioka, D.S. Fullerton, and D.C. Rohrer, Steroids 32, 511 (1978).

¹⁶H.T.A. Cheung, L. Brown, J. Boutagy, and R. Thomas, J. Chem. Soc., Perkin I. 1773 (1981).

¹⁷S.D. Jolad, J.J. Hoffman, and J.R. Cole, *J. Org. Chem.* **46**, 1946 (1981).

18a_M. Karplus, J. Chem. Phys. 30, 11 (1959);
 ^b_R.V. Lemieux, and J.D. Stevens. Can J. Chem.,
 43 2059 (1965); ^c_P.L. Durette, and D. Horton,
 Adv. Carbohydrate Chem. Biochem. 26, 49 (1971).

¹⁹M. Karplus, J. Chem. Phys. 30, 11 (1959).

 20 K. Go, and G. Kartha, *Acta Cryst.* **B36**, 3034 (1980).

²¹K. Go, G. Kartha, and J.P. Chen, *ibid* **B36**, 1811 (1980).

 $^{22}\mathrm{K}.$ Go, and G. Kartha, Crystal Struct. Comm. 11, 279 (1982).

²³K. Go, and G. Kartha, *ibid* **11**, 285 (1982).